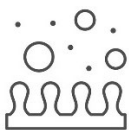
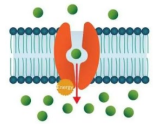




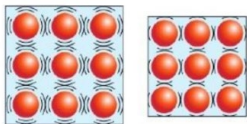
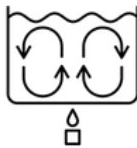
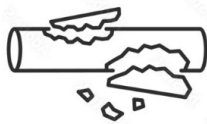

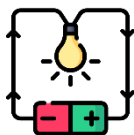
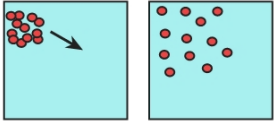





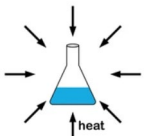
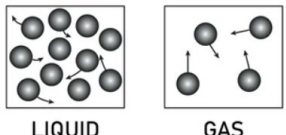

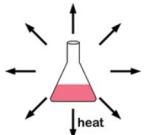
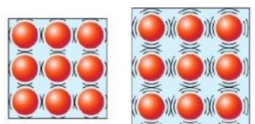






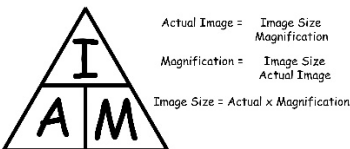





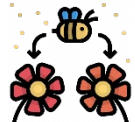


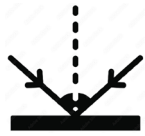

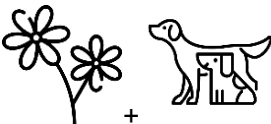
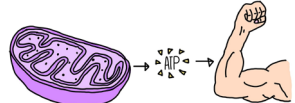

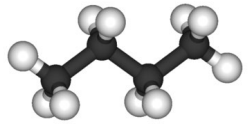
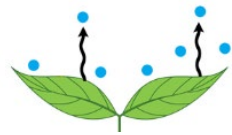
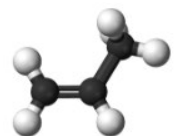


One-word answers: name / identify

absorption		how food molecules move from the small intestine into the bloodstream (through the villi)	
active transport		movement of molecules from a region of lower concentration to a region of higher concentration with the use of energy	
chromatography		method used to separate coloured chemicals or substances	
combustion		any reaction where something (fuel) is burning; needs oxygen, O ₂	
condensation		water vapour losing energy when it cools and turning into liquid	
conduction		transfer of heat through different materials by the vibration of particles	
contraction		gets smaller; spaces between particles change as particles vibrate less and take up less space; particles stay the SAME size	
convection		heat transfer due to movement of particles in a fluid (air or water)	
corrosion		how metals 'break down' (corrode) due to a chemical reaction; usually rusting of iron or steel	
cracking		breaking large high-boiling point hydrocarbon molecules into smaller and more useful molecules	
current (electrical)		flow of charged particles* through a conductor *electrons in metals *ions in electrolysis	

diffusion		movement of anything from a region of higher concentration to a region of lower concentration; can be gas or a liquid	
digestion		breaking down food by mechanical and chemical digestion (enzymes) into substances that can be used by the body	
dispersal (seed)		movement, spread or transport of seeds away from the parent plant	
distillation		separating substances in a liquid mixture; the separation process involves evaporation followed by condensation	
egestion		elimination of undigested food in the form of faeces, the last stage of digestion	
electrolysis		chemical decomposition by passing electric current through a liquid (molten/melted substance) or aqueous solution (dissolved in water)	
endothermic		reaction or process where heat energy is absorbed (taken in); the opposite of exothermic	
evaporation		process of turning from liquid into vapour, at the surface of the liquid	
excretion		process of eliminating waste matter; urine, sweating, exhaling (excretion is NOT same as egestion)	
exothermic		reaction or process where heat energy is released; the opposite of endothermic	
expansion		gets bigger; spaces between particles change as particles vibrate more and take up more space; particles stay the SAME size	
fertilisation		joining of an egg and sperm leading to the formation of a zygote (fertilised egg cell)	

filtration		passing a substance through a filter paper; insoluble residue remains on paper; filtrate collects in a flask	
fractional distillation		separation of a crude oil mixture into fractions based on their boiling points	
galvanising		dipping iron or steel into molten zinc to prevent rusting / corrosion	
germination		plant grows from a seed into a seedling	
ingestion		taking food or liquid into the body	
magnification			
neutralisation		acid + base → salt + water; involves the reaction of H ⁺ ions and OH ⁻ ions to make water, H ₂ O	<p>pH Scale</p> 
osmosis		molecules of water pass through a semipermeable membrane from a less concentrated solution into a more concentrated one	
phagocytosis		ingestion of bacteria or other pathogens by white blood cells (phagocytes)	
photosynthesis		<p>how green plants use sunlight to make glucose (and oxygen) from carbon dioxide and water</p> $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$	
pollination		transfer of pollen from anther to a stigma of a flower to allow fertilisation to occur	
potential difference		difference of electrical potential (voltage) between two points	

radiation		heat energy emitted by matter in the form of electromagnetic waves (infrared)	
reflection		when waves hit a surface or other boundary and bounce away from the surface	
refraction		bending of light as it passes from one transparent substance (medium) into another	
reproduction		production of offspring by a sexual or asexual process	
respiration		production of energy in cells from digested food glucose + oxygen → carbon dioxide + water $C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$	
rusting		formation of red-brown iron(III) oxide on iron; needs iron and oxygen and water	
saturated		saturated hydrocarbons are molecules of C and H with only single C-C bonds	
transpiration		loss of water vapour through the stomata of a leaf	
unsaturated		unsaturated hydrocarbons are molecules of C and H atoms with a double C=C or a triple C≡C bond	
vaccination		treatment with a vaccine to produce immunity to a particular disease / disease causing organism (pathogen)	